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Ionic Compounds

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~~Compounds Ionic~~

~~and Covalent~~

~~Compounds:~~

~~Writing Names~~

~~and Formulas~~

~~Naming Binary~~

~~Ionic Compounds~~

~~With Transition~~

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~~Polyatomic Ions~~

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Equations and  
Net Ionic  
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*Valence  
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Covalent  
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Naming Compounds  
in Chemistry  
Naming Ionic

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(Simple Binary  
Ionic) Tricks

*for Remembering  
Polyatomic Ions*

~~Naming Ionic~~

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Bonding~~

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Rules for Naming  
Covalent

Compounds *Naming  
Covalent*

*Compounds*

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Chemical Naming:

Covalent

Compounds Name

Covalent

Compounds Easily

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**Compounds**

Writing Ionic

Formulas:

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Naming Chemical  
Compounds

Determine  
whether the  
following are  
ionic or  
covalent, and  
then

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*ionic and covalent  
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Compound Naming

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Naming Covalent

Compounds

Solutions. Write

the formulas for

the following

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compounds: 1)

antimony

tribromide

$\text{SbBr}_3$ . 2)

hexaboron

silicide  $\text{B}_6\text{Si}$  3)

chlorine dioxide

$\text{ClO}_2$ . 4)

hydrogen iodide

$\text{HI}$  5) iodine

pentafluoride

$\text{IF}_5$ . 6)

dinitrogen

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Naming

trioxide N<sub>2</sub>O<sub>3</sub>.

Compounds

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Nomenclature

Worksheet 3 –

Covalent

(Molecular)

Compounds Part

A: Name the

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Naming

following

covalent

compounds. 1. CO

Carbon monoxide

. 2. CO<sub>2</sub> Carbon

dioxide. 3. N<sub>2</sub>O<sub>3</sub>

Dinitrogen

trioxide. 4. N<sub>2</sub>

monit. 5. NP

Nitrogen

monophosphide.

6. SCl<sub>2</sub> Sulfur

dichloride. 7.

P<sub>2</sub>O<sub>5</sub>

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Naming

Diphosphorous  
pentoxide. 8.

$\text{NBr}_3$  Nitrogen  
tribromide. 9.

$\text{CI}_4$  Carbon  
tetraiodide. 10.

*Nomenclature*  
*Worksheet 3 –*  
*Covalent*  
*(Molecular)*

*Compounds*  
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covalent ionic  
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compounds,  
Covalent,  
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Naming Covalent

Compounds (Type

III) Worksheet.

Write the

formulas for the

following

covalent

compounds: 1)

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Naming

nitrogen  
tribromide \_\_\_\_\_

\_\_\_\_\_

Answer Sheet

2) hexaboron  
silicide \_\_\_\_\_

\_\_\_\_\_ . 3)  
chlorine dioxide

\_\_\_\_\_

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compounds,

Covalent  
Compounds

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properties of  
the following



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Steps to Naming  
Covalent  
Compounds.

First, identify  
the elements  
present. Second,  
look at the  
subscript of

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## Naming

each element to determine which prefix to use.

(If an element does not have a prefix, assume that the subscript is "1." Third, apply the above naming scheme.

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